

Potassium Reagent Kit

IVD

Colorimetric Test

PRODUCT CODE CE005

INTENDED USE

This reagent kit is intended for the "In Vitro "quantitative determination of Potassium in Serum

CLINICAL SIGNIFICANCE

Potassium is an electrolyte that is vital to cell metabolism. It helps transport nutrients into cells and removes waste products out of cells. It is also important in muscle function, helping to transmit messages between nerves and muscles, Elevated potassium levels (hyperkalemia) are often associated with renal failure, dehydration shock or adrenal insufficiency. Decreased potassium levels (hypokalemia) are associated with malnutrition, negative nitrogen balance, gastrointestinal fluid losses and hyperactivity of the adrenal cortex, this test measures the amount of potassium in the blood

PRINCIPLE

Potassium reacts with sodium tetra phenol boron in a specially prepared buffer to form a colloidal suspension. The amount of the turbidity produced is directly proportional to the concentration of potassium in the sample.

REAGENT COMPOSITION

Reagent 1: Potassium Reagent

Reagent 2: Potassium Standard 5 mEq/L

REAGENT PREPARATION

The reagent and standard are ready to use.

STORAGE AND STABILITY

The reagents and standard are stable up to the stated expiry date when stored at 15-30 $^{\circ}$ C.

SPECIMEN

Serum free from hemolysis

Separate serum from the clot as soon as possible as Potassium may leach from the RBC's which can elevate potassium level.

NORMAL RANGE

Serum/Plasma: 3.5 - 5.5 mEq/L.

It is recommended that each laboratory establish its own normal range representing its patient population.

ASSAY

Wavelength 630 nm
reaction type Endpoint
Cuvette 1 cm light path
Temperature Room temp.
Measurement Against reagent blank

PROCEDURE

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Pipette in to cuvettes	Blank	Standard	Sample
Potassium Reagent	1000 μL	1000 μL	1000 μL
Standard		20 μL	
Sample			20 μL

Mix and incubate for 5 min at room temp. And measure the absorbance of the sample (As) and the standard (A.std) against the reagent blank.

CALCULATION

Concentration of Potassium (mEq/L). = $\frac{\Delta A \text{ sample}}{\Delta A \text{ standard}} X 5 \text{ (Std. conc.)}$

LINEARITY

This procedure is linear up to 7 mEq/L. If values exceed his limit dilute the sample with distilled water and multiply results with proper dilution factor.

NOTE

As potassium is a very widely distributed ion, care should be taken to avoid any contamination. All glass wares being used for the test should first be rinsed with 1% or 0.1 N HNO3 and then with good quality deionized water before use.

QUALITY CONTROL

Control serum of known concentrations should be analyzed with each run

SYMBOL ON LABELS

Symbols	Signify	Symbols	Signify
REF	Catalogue Number	SIZE	Pack Size
Ω	Expiry Date	VOL	Volume
*	Storage Condition	LOT	Lot Number
[]i	Instruction for Use	IVD	In Vitro Diagnostics
\mathbb{A}	Manufacturing Date	***	Manufacturer
Σ	Number of Tests	2	For Single Use Only
EC REP	EC Representative	Œ	European conformity

BIBILOGRAPHY

1-Tietz, N.W., Fundamentals of Clinical Chemistry, W.b

Saunders Co. Phila, P.A. p. 874.

2-enry R.F., et, al, Clinical Chemistry Principles and Te2-chnics. 2ndE

d, H arper and R ow, Harper and Row, Hargersein, M.D. (1974)

3- Maruna RFL., Clin Chem. Acta. 2:581, (1958)

4- Trinder, P:Analyst, 76:596, (1951)

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